

The Second Coenzyme Q₁ Binding Site of Bovine Heart NADH: Coenzyme Q Oxidoreductase

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The rotenone sensitivity of bovine heart NADH: coenzyme Q oxidoreductase (Complex I) depends significantly on coenzyme Q₁ concentration. The rotenone-insensitive Complex I reaction in Q₁ concentration range above 300 μM indicates an ordered sequential mechanism with Q₁ and reduced Q₁ (Q₁H₂) as the initial substrate to bind to the enzyme and the last product to be released from the enzyme product complex, respectively. This is the case in the rotenone-sensitive reaction although both K_m and V_{max} values of the rotenone-insensitive reaction for Q₁ are significantly higher than those of the rotenone-sensitive reaction (Nakashima *et al.*, 2002, *J. Bioenerg. Biomemb.* **34**, 11–19). This rigorous control mechanism between the nucleotide and ubiquinone binding sites strongly suggests that the rotenone-insensitive reaction is also physiologically relevant.

KEY WORDS: NADH: coenzyme Q oxidoreductase; Complex I; membrane protein; steady state kinetics; ordered sequential mechanism; mitochondrial respiration; coenzyme Q.

INTRODUCTION

NADH: coenzyme Q oxidoreductase (EC 1.6.99.3) (Complex I²) is at the entrance of the mitochondrial respiratory chain (Hatefi, 1985; Walker, 1992; Weiss *et al.*, 1991) and catalyzes the reduction of coenzyme Q (Q) via a mechanism involving NADH coupled to the translocation of protons to the intermembrane space in a ratio of $4H^+/2e^-$ (Di Virgilio and Azzone, 1982; Esposti and Ghelli, 1994; Scholes and Hinkle, 1984; Weiss and Friedrich, 1991; Wikstöm, 1984). Complex I is an extremely large membrane protein complex with a flavin (FMN) (Hatefi, 1985; Walker, 1992; Weiss *et al.*, 1991) and several iron–sulfur clusters acting as redox active catalytic centers (Ohnishi, 1998; Ohnishi *et al.*, 1985). Due to difficulties encountered in the isolation of the large membrane-bound protein, both the structure and the reaction mechanism of this enzyme are poorly understood.

In our previous paper, we reported a steady state kinetic study on the rotenone-sensitive reaction which indicates an ordered sequential mechanism with the order of the substrate binding and product release as follows: Q₁—NADH—NAD⁺—Q₁H₂ (Nakashima *et al.*, 2002). The mechanism suggests a long-range interaction between the NADH binding site and the Q₁ binding site. An increase in the rotenone-insensitive enzyme activity is observed with an increase in Q₁ concentration above the saturating Q₁ concentration for the rotenone-sensitive reaction. Thus we investigated the steady state kinetics of the rotenone-insensitive reaction, for elucidation of the function of this enzyme, particularly for understanding the mechanism of the rotenone inhibition. A complete steady state kinetic analysis of the reaction (Cleland, 1977) shows an ordered sequential mechanism with Q₁ and Q₁H₂ as the initial substrate to bind to the enzyme and the last product released, respectively. The rigorous control in the mechanism suggests that the rotenone-insensitive reaction is also physiologically relevant.

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Key to abbreviations: Complex I, NADH: coenzyme Q oxidoreductase; Q_n, coenzyme Q (ubiquinone) with *n* isoprenoid units; Q₁H₂, reduced coenzyme Q₁; NMR, nuclear magnetic resonance.

EXPERIMENTAL PROCEDURES

Materials and enzyme reaction assay methods are as described in the previous paper (Nakashima *et al.*, in press).

The NMR relaxation time (T_1) of Q_1 was measured with a JEOL Excalibur 400 spectrometer. NMR samples were measured in D_2O in the presence of dodecyl maltoside, ethanol, and phosphate buffer at the same concentrations as in the enzyme reaction mixture. Deuterated ethanol was used to remove the background due to ethanol $-OH$. A PW_2 (180° pulse) of $34 \mu s$ and PW_1 (90° pulse) of $17 \mu s$ were applied sequentially at a fixed 10-s interval while the pulse sequence repetition interval ranged from 0.1 to 5 s. Data were averaged over 40–400 runs depending on the concentration of Q_1 .

RESULTS

Dependence of Rotenone-Sensitivity on Q_1 Concentration

At a saturating concentration of rotenone significantly higher than the enzyme concentration (Nakashima *et al.*, in press), the rotenone inhibition was influenced strongly by Q_1 concentration. As shown in Fig. 1, the fractional inhibition of rotenone at 150 nM in the presence of $10 \mu M$ NADH decreased significantly with Q_1 concentration above $50 \mu M$. At a Q_1 concentration below $20 \mu M$, rotenone inhibition did not exceed 85%. On the other hand, above $400 \mu M$, the rotenone-sensitive activity was much lower than that of the total activity (lower than 7%). In the high Q_1 concentration range, the saturation level of rotenone concentration was essentially the same as those in the Q_1 concentration range below $50 \mu M$. Increasing the rotenone concentration above 150 nM up to $1 \mu M$ did not increase the rotenone inhibition in the presence of high concentrations of Q_1 .

In the Q_1 concentration range for the previous and present experimental conditions (i.e. below $600 \mu M$), NADH oxidation in the initial linear portion of the reaction was stoichiometric to the reduction of Q_1 , monitored by the absorbance change at 290 nM, as described in the previous paper (Nakashima *et al.*, in press). The molar ratio of NADH oxidized to Q_1 reduced did not deviate from unity more than 0.05.

Steady State Kinetic Analysis

Steady State properties were examined mainly above $400 \mu M$ Q_1 where the rotenone-insensitive reaction accounted for more than 90% of the total activity. As shown

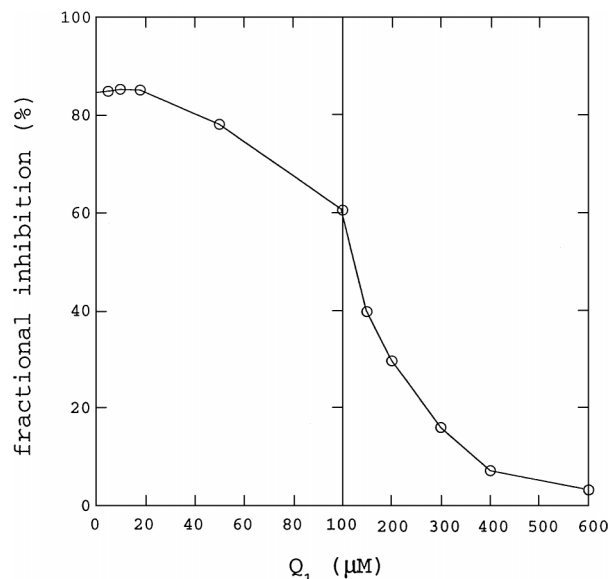


Fig. 1. Effect of Q_1 concentration on the rotenone sensitivity of Complex I. Fractional rotenone-inhibition values (percent decrease in the enzyme activity) of Complex I preincubated with an excess amount of rotenone as described in the text are plotted against the Q_1 concentration in the reaction mixture. The final concentrations of Complex I, rotenone, and NADH in the reaction mixture were $1.84 \mu g/mL$, $150 nM$, and $10 \mu M$, respectively, in $0.1 M$ potassium phosphate buffer, pH 8.0, containing 0.1% dodecyl maltoside.

in Fig. 2, simple rectangular hyperbolic curves were obtained at three fixed Q_1 concentrations (400 , 500 , and $600 \mu M$). The statistically calculated values for V_{max} and K_m are given in Table I. The K_m values are identical within experimental error, indicating that the three double reciprocal plots ($1/v_0$ versus $1/NADH$) intersect at the abscissa as shown in the inset. The solid lines in the main panel are rectangular hyperbolic curves constructed using the V_{max} and K_m values given in Table I. No systematic deviation is detectable from the theoretical curves. The results indicate a sequential mechanism for substrate binding and product release which includes an enzyme— Q_1 —NADH ternary complex.

Product inhibition was examined for determination of the order of the substrate binding and the product release for all substrate–product combinations, as given in Fig. 3. A set of intersecting double reciprocal plots was obtained for each combination. Only the Q_1 – Q_1H_2 combination gives the intersecting point at the ordinate showing competitive inhibition of Q_1H_2 against Q_1 . Because of the large K_m values for Q_1 , curves in Fig. 3(C) and (D) appear to be linear. However, Q_1 concentration ranges for these data points are not much lower than K_m values. In fact, the v_0 values on each solid line are not proportional to Q_1 concentration. Thus, it is reasonable to assume a

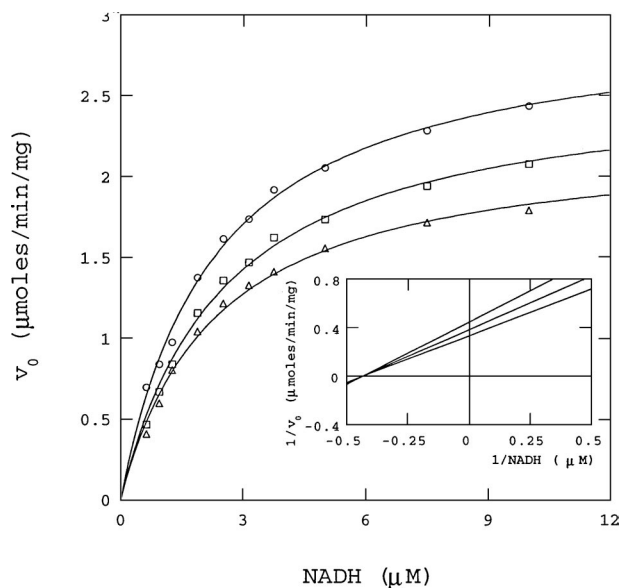


Fig. 2. NADH-dependent oxidation rates at various fixed concentrations of Q₁ in the absence of products. The Q₁ concentrations were 600 μM (circle), 500 μM (square), and 400 μM (triangle). The enzyme concentration was 10.25 μg/mL in 0.1 M potassium phosphate buffer, pH 8.0, containing 0.1% dodecyl maltoside. Solid curves were obtained by fitting the data to the Michaelis–Menten equation. The double reciprocal plots of the NADH-dependent rates obtained by the V_{\max} and K_m values, which give the rectangular hyperbolic curves in the main panels, are shown in the insets.

rectangular hyperbolic relationship (i.e. Michaelis–Menten mechanism) for each dataset. On the basis of the standard errors for K_m and V_{\max} , it is statistically reasonable to conclude the competitive and noncompetitive inhi-

tion by Q₁H₂ and NAD⁺ respectively (Panels D and C). Experimental results obtained at Q₁ concentration below 400–300 μM are also included in Fig. 3(C) and (D). The data points above 400 μM Q₁ provide rectangular hyperbolic curves, each identical to the one given in the figures within the experimental error. This shows that the influence of the rotenone-sensitive reaction to the rotenone-insensitive reaction is negligible below 400 μM down to 300 μM Q₁.

This steady state kinetic analysis indicates an ordered sequential mechanism with the order of substrate binding and product release as Q₁—NADH—NAD⁺—Q₁H₂, as given in Scheme 1 (Cleland, 1977).

Examination of a Possible Secondary Effect on the Enzyme Preparation due to the High Concentration of Q₁

The Q₁ concentration range used in the present work is much higher (up to 600 μM) than those in the previous reports (below 120 μM) (Esposti *et al.*, 1996; Fato *et al.*, 1996). It might be expected that high concentrations of Q₁ (which is at least partly hydrophobic) might cause denaturation of the purified Complex I which is not very stable. Thus, we examined the effect of high concentrations of Q₁ on enzyme activity. Complex I (10.25 μg/mL) was preincubated with 600 μM Q₁ at 20°C for 30 s before addition of the 0.2 mL portion of the preincubation mixture to 1.8 mL of the reaction mixture containing 11.1 μM NADH. The addition of the enzyme solution yields 60 μM Q₁ and 10 μM NADH in the initial state of

Table I. Kinetic Parameters for NADH-Q₁ Reductase

Varying substrate	Fixed substrate (μM)	Product (μM)	V_{\max} (μmoles/min/mg)	K_m (μM)
NADH	Q ₁ 400	0	2.240 (0.060)	2.275 (0.159)
	500	0	2.613 (0.064)	2.494 (0.154)
	600	0	2.988 (0.065)	2.274 (0.128)
NADH	Q ₁ 400	NAD ⁺ 0	2.240 (0.060)	2.275 (0.159)
		100	1.545 (0.024)	1.721 (0.078)
		200	1.406 (0.020)	1.580 (0.069)
		400	1.025 (0.013)	1.535 (0.060)
NADH	Q ₁ 400	Q ₁ H ₂ 0	2.240 (0.060)	2.275 (0.159)
		200	1.707 (0.023)	1.633 (0.065)
		300	1.414 (0.025)	1.471 (0.082)
Q ₁	NADH 6	NAD ⁺ 0	3.724 (0.149)	542.7 (39.6)
		100	2.913 (0.102)	534.9 (34.4)
		200	2.473 (0.092)	519.2 (36.1)
		400	1.980 (0.069)	504.3 (43.3)
Q ₁	NADH 6	Q ₁ H ₂ 0	3.724 (0.149)	542.7 (39.6)
		200	3.805 (0.200)	696.5 (60.3)
		300	3.792 (0.162)	804.3 (53.6)

Note. Values given in parentheses are standard errors.

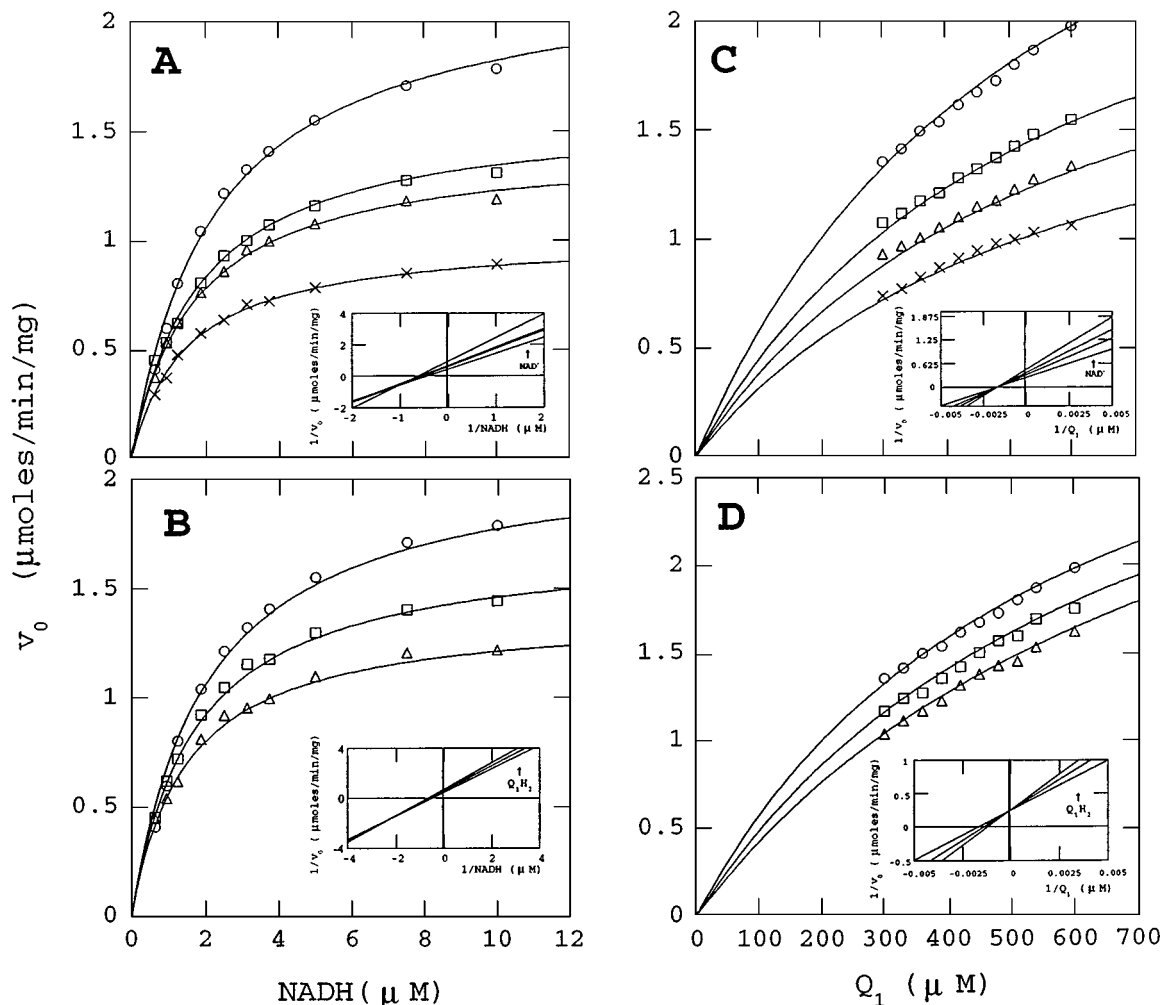


Fig. 3. Product inhibition for the NADH:Q₁ reductase reaction at Q₁ concentrations above 400 μM . Solid curves were determined by fitting the data to the Michaelis–Menten equation. The double reciprocal lines of the solid curves are shown in the insets. (A) NAD⁺ concentrations are 0 μM (circle), 100 μM (square), 200 μM (triangle), and 400 μM (cross) with varying concentrations of NADH at a fixed Q₁ concentration of 400 μM . (B) Q₁H₂ concentrations are 0 μM (circle), 200 μM (square), and 300 μM (triangle) with varying concentrations of NADH at a fixed Q₁ concentration of 400 μM . (C) NAD⁺ concentrations are 0 μM (circle), 100 μM (square), 200 μM (triangle), and 400 μM (cross) with varying concentrations of Q₁ at a fixed NADH concentration of 6 μM . (D) Q₁H₂ concentrations are 0 μM (circle), 200 μM (square), and 300 μM (triangle) with varying concentrations of Q₁ at a fixed NADH concentration of 6 μM . In C and D, data points in the presence of Q₁ below 400 μM are also included. The enzyme concentration was 10.25 $\mu\text{g}/\text{mL}$ in 0.1 M potassium phosphate buffer, pH 8.0, containing 0.1% dodecyl maltoside.

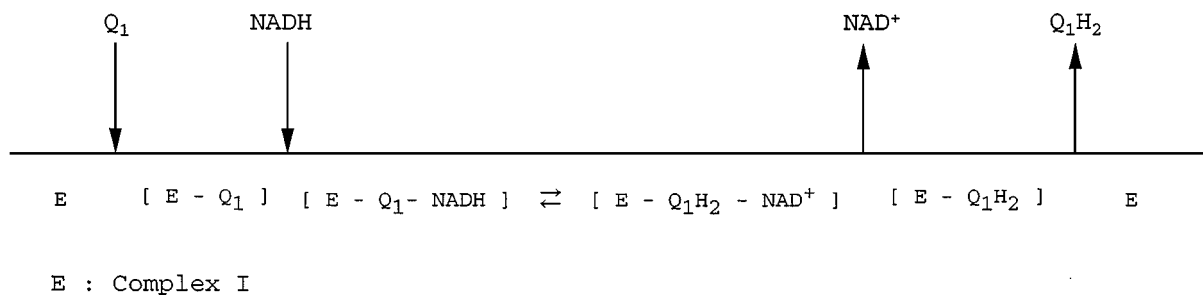
the reaction mixture. The specific activity was identical to those determined without preincubation and with preincubation in the absence of Q₁, indicating that Q₁ at 600 μM does not affect the enzyme activity of our Complex I preparation.

As described in the previous paper (Nakashima *et al.*, in press), rotenone inhibition is not instantaneous. Thus, an initial burst in decrease of the slope of the NADH oxidation is observed during the initial 60 s. The initial burst is removed by preincubation of the enzyme with rotenone for 30 s at 20°C. No initial burst was detectable in the NADH

oxidation when the enzyme preparation was preincubated with rotenone in the presence of 600 μM Q₁. This result shows that 600 μM Q₁ in the preincubation mixture does not inhibit the binding of rotenone to the enzyme.

NMR Relaxation of Q₁

It might be expected that high concentrations of Q₁ could form micelles or other aggregates in aqueous solution that could cause activity artifacts. However, the



Scheme 1.

structure of Q₁ indicates that it is unlikely to form micelles and is moderately hydrophilic. However, in order to show the homogeneous dispersion of Q₁ in aqueous solution most directly, we measured the NMR relaxation time (T_1) of protons of three methyl groups of Q₁, one at position 5 in the quinone ring and two in the isoprenoid group. These methyl protons exhibit three clear peaks at 1.80, 1.55, and 1.48 ppm for 600 μ M Q₁ in 100 mM potassium phosphate buffer, pH 8.0, containing 1% deuterated ethanol and 0.1% dodecyl maltoside. The T_1 values of NMR relaxation for these signals were 0.77, 1.08, and 0.89 s, respectively. The T_1 values and peak positions of these signals were not influenced by Q₁ concentrations between 600 and 100 μ M. Similarly ethanol did not significantly affect these NMR parameters. Furthermore, essentially identical NMR parameters were obtained for these three methyl signals in deuterated chloroform. These results indicate that Q₁ at concentrations below 600 μ M is homogeneously dispersed in the reaction mixture without forming micelles or aggregates.

DISCUSSION

The Rotenone-Sensitive and Rotenone-Insensitive Reactions

Rotenone sensitivity was decreased with Q₁ concentration and only about 3% of the total activity was inhibited by addition of an excess amount of rotenone in the presence of 600 μ M Q₁. On the other hand, about 80% of the total activity was abolished by rotenone at Q₁ concentrations below 50 μ M. These results suggest that this enzyme catalyzes two types of reactions, a rotenone-sensitive reaction and a rotenone-insensitive reaction. The latter appears at high Q₁ concentrations because of the large K_m values for Q₁. The existence of the rotenone-insensitive reaction was also demonstrated by the enzyme activity increase in the Q₁ concentration range where the rotenone-sensitive reaction (K_m of 12 μ M for Q₁) was fully saturated

with Q₁ (data not shown). In contrast to the NAD⁺ inhibition against NADH for the rotenone-sensitive reaction (Nakashima *et al.*, in press), NAD⁺ decreases the apparent K_m value for NADH (Fig. 3(A) and Table I). In other words, the reactivities of NADH and NAD⁺ to the enzyme are influenced by the Q₁ binding states, indicating a long-range interaction between the pyridine nucleotide and ubiquinone binding sites.

As described in the previous paper (Nakashima *et al.*, in press), this enzyme has only a single site for rotenone binding. The rate of the rotenone-insensitive reaction should decrease with a decrease in the concentration of Q₁ asymptotically to zero. However, as shown in Fig. 1, the fractional rotenone inhibition approached to about 85% with a decrease in the concentration of Q₁. This result suggests that rotenone cannot inhibit the enzyme activity completely even at low Q₁ concentrations. Thus, it is likely that the rotenone-insensitive activity at extremely low Q₁ concentrations compared with K_m value of the rotenone-insensitive reaction for Q₁ is due to residual activity of the rotenone-bound site, not due to the rotenone-insensitive reaction observed at high Q₁ concentrations.

The integrity of purified preparations of Complex I has been evaluated by investigation of the rotenone sensitivity of the enzyme activity (Hatefi and Stempel, 1969). The rotenone-insensitive activity observed at high Q₁ concentration is unlikely to be due to denaturation of the purified preparation since Q₁ at 600 μ M did not cause denaturation of the Complex I preparation. Additionally, aggregates or micelles (which could affect the kinetics of the enzyme) were not detectable even at the highest concentration of Q₁ as revealed by the NMR relaxation measurement. These results indicate that the rotenone-insensitive activity observed here is not due to denaturation of the enzyme. Thus, for an accurate evaluation of the integrity of purified Complex I, the rotenone sensitivity should be examined at low Q₁ concentration where the rotenone-insensitive reaction is negligible (lower than 20 μ M).

Physiological Significance of the Rotenone-Insensitive Reaction

The high K_m value for Q_1 in the rotenone-insensitive reaction seems consistent with nonphysiological Q_1 binding on the hydrophilic surface of the enzyme, as previously suggested (Ragan, 1978). However, the overall Q_{10} concentration in mitochondrial membrane is definitely within the millimolar range. It has been reported that Q_{10} content is 1.06–0.43% of the total lipid of bovine mitochondria (Fleischer *et al.*, 1967). The overall concentration of Q_{10} (1.06%) in bovine heart mitochondrial lipid phase corresponds to 11.6 mM (Fleischer *et al.*, 1967). Furthermore the effective concentration of Q_{10} to the enzyme could be higher than the overall concentration since the movement of Q_{10} is restricted significantly in the mitochondrial inner membrane. Thus, the rotenone-insensitive Q_1 binding site with the large K_m values for Q_1 (but lower than mM) is likely to be active in the enzyme reaction in the mitochondrial inner membrane. Furthermore, the rotenone-insensitive reaction indicates that a rigorous long-range control exists between the NADH binding site and the rotenone-insensitive Q_1 binding site in a manner similar to the rotenone-sensitive reaction. Such a rigorous control mechanism also strongly suggests that the rotenone-insensitive reaction is physiologically relevant.

On the other hand, it is well known that rotenone inhibits almost completely the Complex I reaction in intact mitochondrial innermembrane (Hatefi, 1985) in which Q_{10} concentration is about 10 mM (Fleischer *et al.*, 1967). A possible interpretation for the apparent discrepancy is as follows: the second ubiquinone (Q_{10}) binding site is composed of two subsites for the quinone moiety near the extramembrane surface of Complex I and for the isoprenoide chain moiety in the transmembrane hydrophobic surface. And rotenone binds only to the latter subsite. Rotenone binding to the latter subsite is sufficient for a complete inhibition of Q_{10} binding to the site in the mitochondrial membrane. However, the short and flexible isoprenoide tail of Q_1 is likely to have an additional binding site different from the isoprenoide subsite for Q_{10} , so that rotenone cannot inhibit Q_1 binding. Thus, NADH— Q_1 oxidoreductase reaction in the present enzyme assay system at high Q_1 concentration is rotenone-insensitive, while NADH— Q_{10} oxidoreductase reaction in the mitochondrial innermembrane is rotenone-sensitive. On the other hand, the rotenone-sensitive reaction in the present

assay system at low Q_1 concentration is inhibited by binding of rotenone to the quinone binding site on the transmembrane surface. Thus, the reaction is sensitive to the rotenone in the isolated enzyme reaction system as well as in the intact mitochondrial membrane. Various alternative interpretations may be possible at present. However, as discussed above, it is unlikely that the rotenone-insensitive activity in the present assay system for the purified enzyme preparation is provided by some denatured enzyme fractions in the enzyme preparation, if any. The structural information at high resolution for the Q sites is required for solving the rotenone-inhibition mechanism.

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